



Type II van der Waals Heterostructures of SiC/Al₂OS monolayers for Photocatalysis and Optoelectronics Applications

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Background

- Global energy demands and consumption have increased over the past few years due to a significant increase in industrialization, urbanization and population.¹
- The creation of photocatalytic hydrogen through water splitting under light irradiation is receiving great attention as an alternative energy to replace fossil fuels.²
- Silicon carbide (SiC) has emerged as a photocatalyst material for effective H₂ evolution due to its electronic, structural, mechanical and optical properties.¹
- 2D Janus Al₂OS monolayer has been theoretically proved that a strong visible light absorption, desirable bandgap that can help improve hydrogen evolution reaction.³

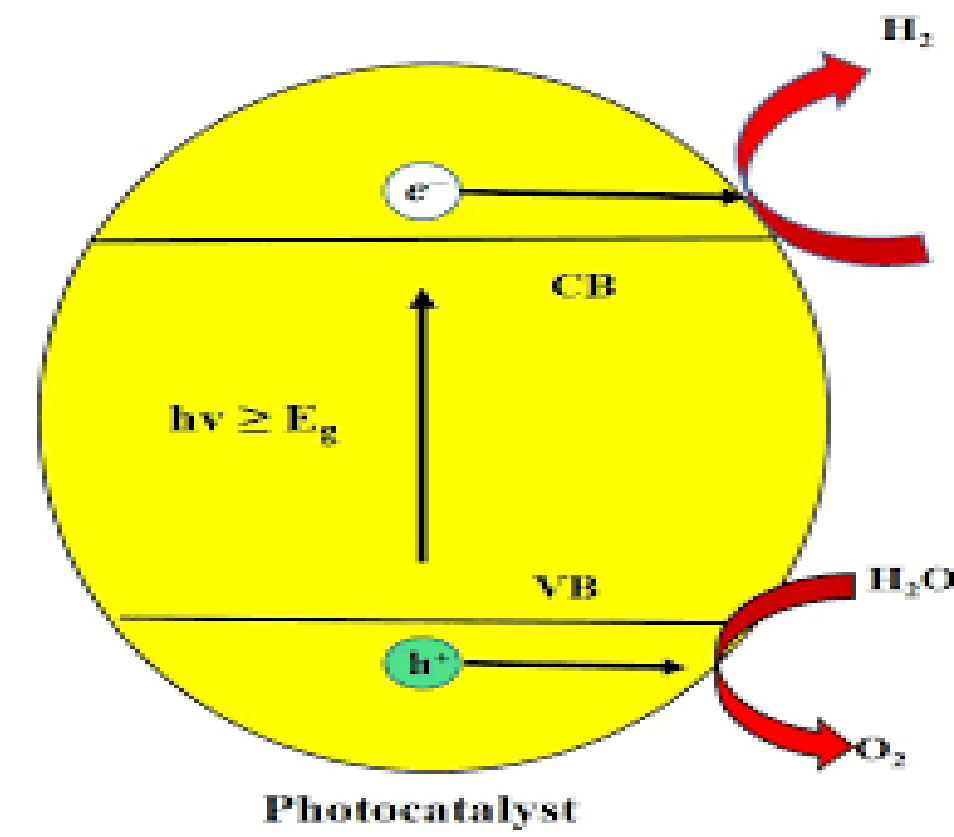


Figure 1 : Mechanism of photocatalytic water splitting.

Aim and Objectives

➤ Aim

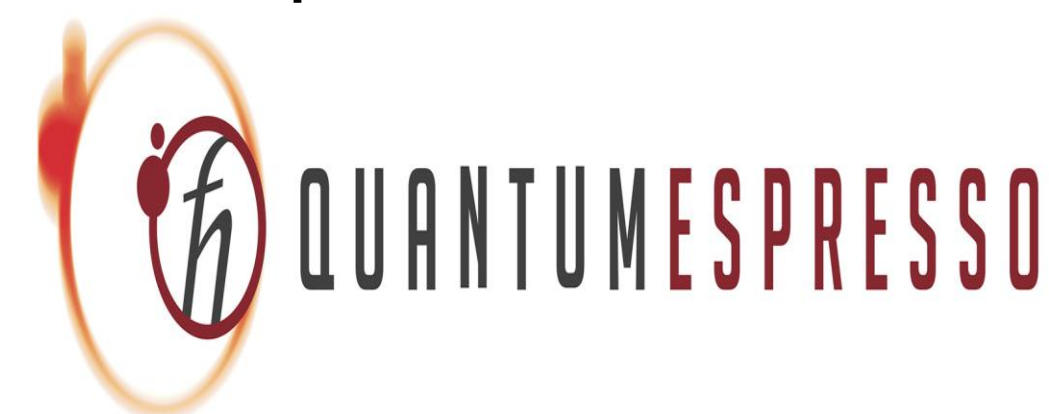
- To investigate the electronic properties of van der Waals heterostructure of SiC and Al₂OS monolayers for photocatalytic and optoelectronic applications.

➤ Objectives

- To evaluate the photocatalytic potential by comparing the band edges with redox potential of water.
- To enhance the charge separation and transfer capabilities of photocatalytic materials.

Computational Details

- Solving the Kohn-Sham equation in DFT with plane wave basis set. The Quantum ESPRESSO code was used to perform DFT calculations. Vacuum thickness 30 Å
- The **cut off energy** was set as **45 Ry** and the Monkhorst-Pack **k point mesh** of **6 × 6 × 1** and **15 × 15 × 1** was used for all the optimization processes and electronic properties calculations, respectively. Heyd-Scuseria-Ernzerhof (HSE06) hybrid functional was used to accurately predict the optical and electronic properties.
- Norm conserving Vanderbilt pseudopotentials to account for the electron-ion interaction
- The interlayer vdW interactions and dipole were treated using the dispersion corrections of Grimme DFT - D3 method.



Results and Discussion

➤ Structure Geometry

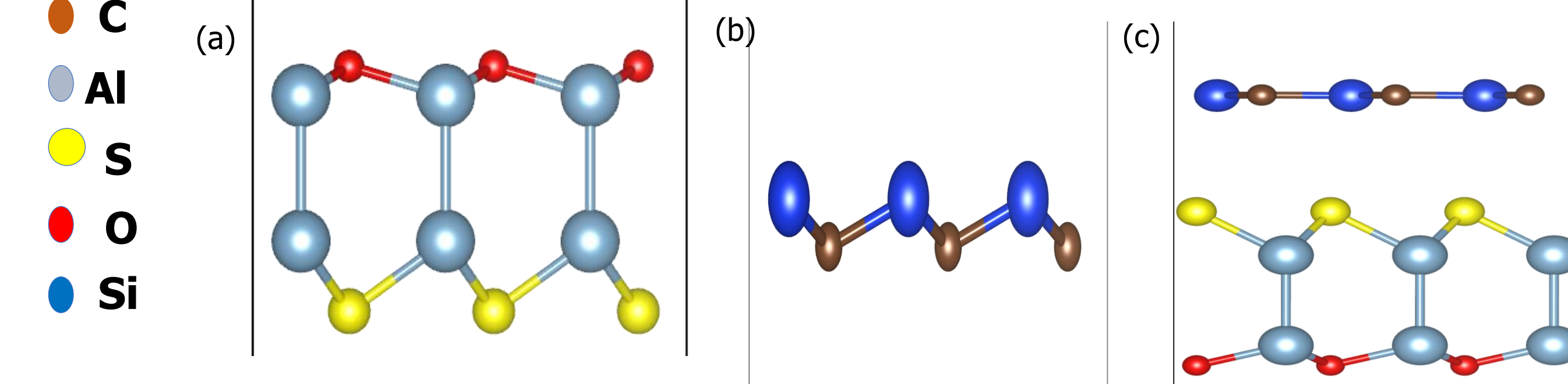


Figure 2 : Optimized side view of (a) Al₂OS monolayer (b) SiC monolayer and (c) The side and top view of SiC/Al₂OS heterostructure

Table 1: Lattice parameters ($a=b$) of SiC and Al₂OS monolayers and SiC/Al₂OS heterostructure.

	($a=b$) (Å)	Experimental (Å)	Lattice mismatch (%)
Al ₂ OS	3.22	3.28	-
SiC	3.07	3.11	-
SiC/Al ₂ OS	3.17	-	2.39

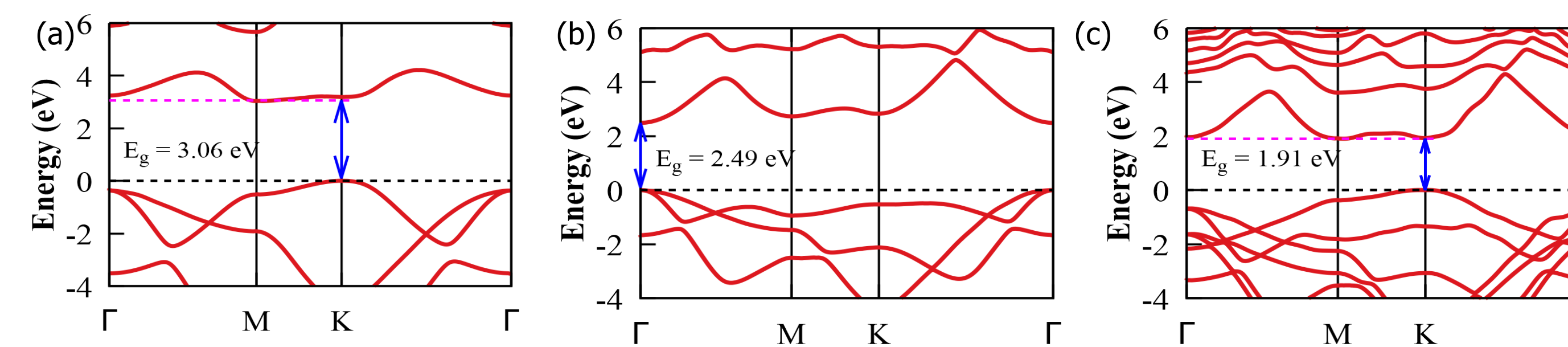


Figure 3 : The calculated band structure of (a) SiC (b) Al₂OS and (c) SiC/Al₂OS heterostructure

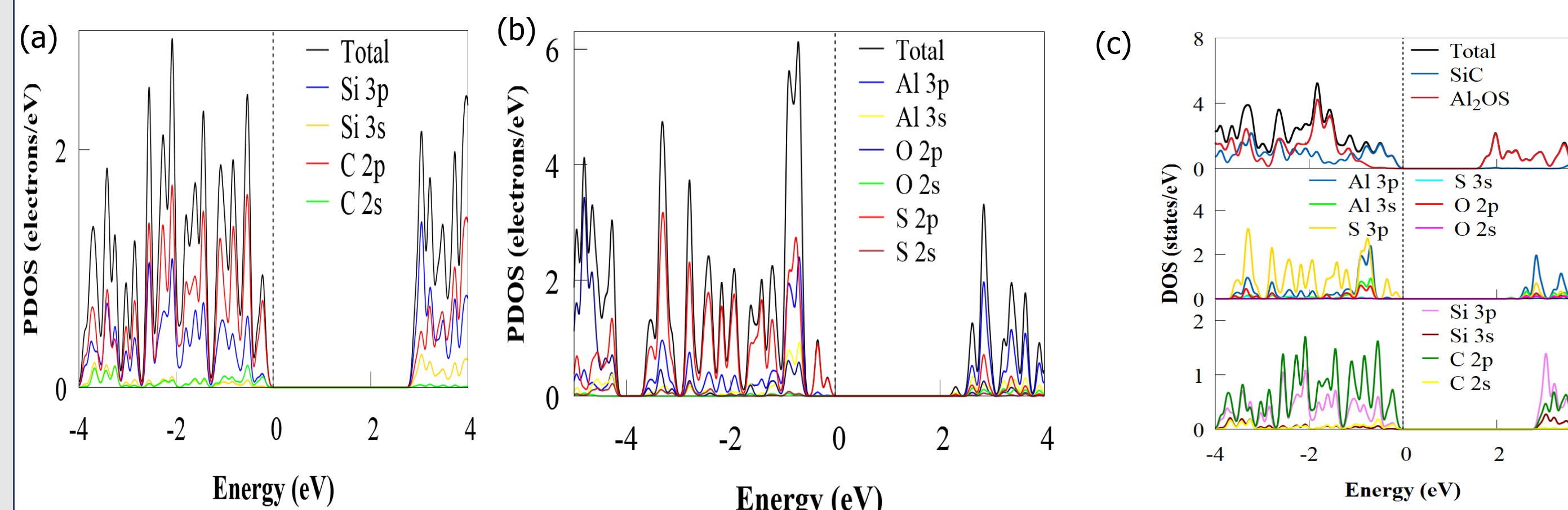


Figure 4 : The calculated projected density of state of (a) SiC (b) Al₂OS and (c) SiC/Al₂OS heterostructure

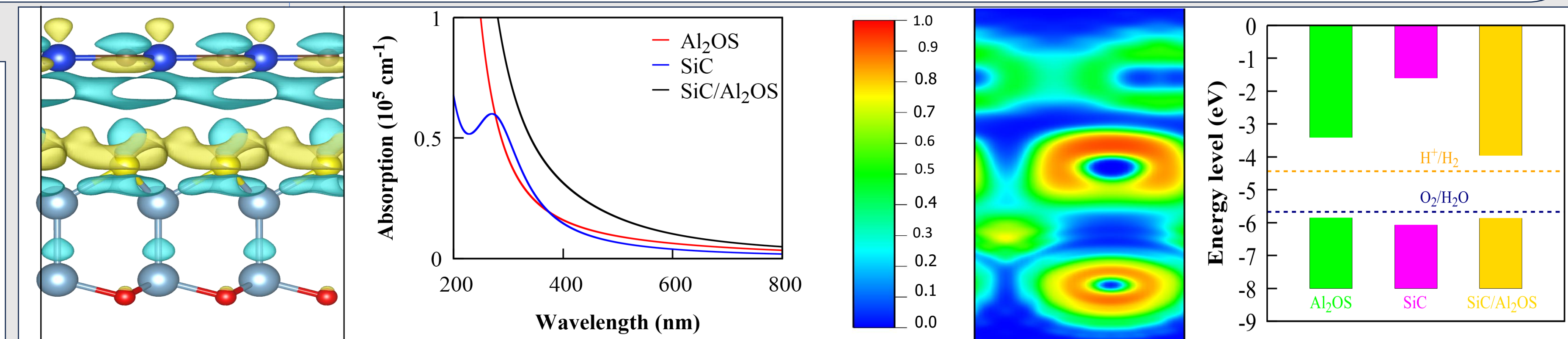


Figure 5: (a) The charge density difference of SiC/Al₂OS heterostructure with the blue and yellow being charge accumulation and depletion, respectively (b) The calculated light absorption spectra of SiC/Al₂OS heterostructure and the monolayers. (c) Electron localization function of SiC/Al₂OS heterostructure and (d) band alignment for the SiC/Al₂OS heterostructure

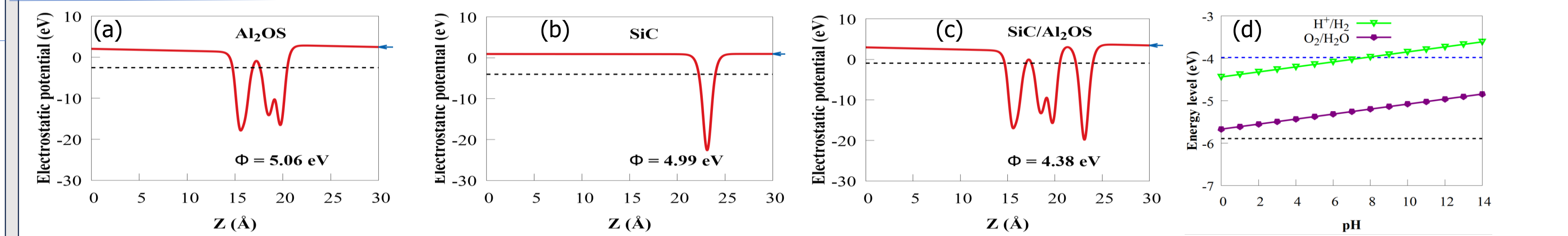


Figure 6: Work function of (a) SiC monolayer (b) Al₂OS monolayer (c) SiC/Al₂OS heterostructure. (d) Band edge potentials against a normal hydrogen electrode at pH = 0 of SiC/Al₂OS vdWHs.

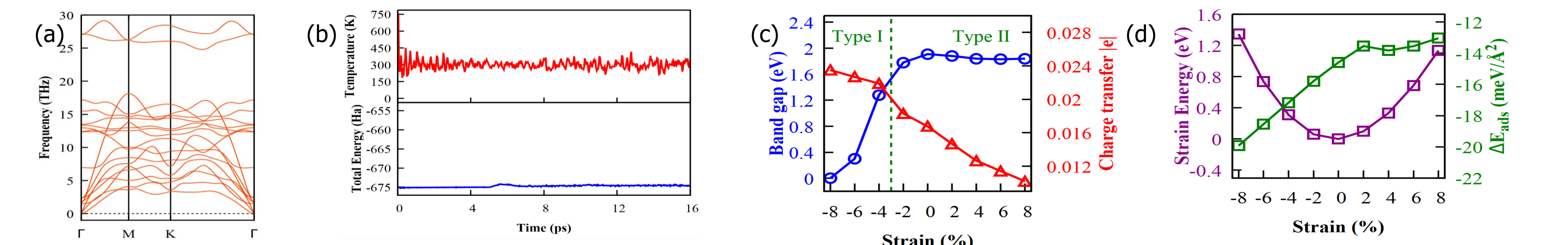


Figure 7 : Snapshots of (a) Phonon dispersion of SiC/Al₂OS vdWHs (b) molecular dynamics at 300K (c) Variation of the interfacial binding energy and the strain energy, and (d) changes in the charge transfer and the band gap energy

Conclusion

- SiC/Al₂OS heterostructure exhibits a direct bandgap of 1.91 eV with a type-II band alignment, which can effectively separate and transfer charge carriers, thereby enhancing the photocatalytic activity.

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- Opoku, F., Aniagyei, A., Akoto, O., Kwaansa-Ansah, E. E., Asare-Donkor, N. K., & Adimado, A. A. (2023). Investigation of 2D Janus Al₂OS/Ga₂SSe van der Waals heterojunction as next-generation thermoelectric and photocatalytic devices. *Next Materials*, 1(4), 100042.