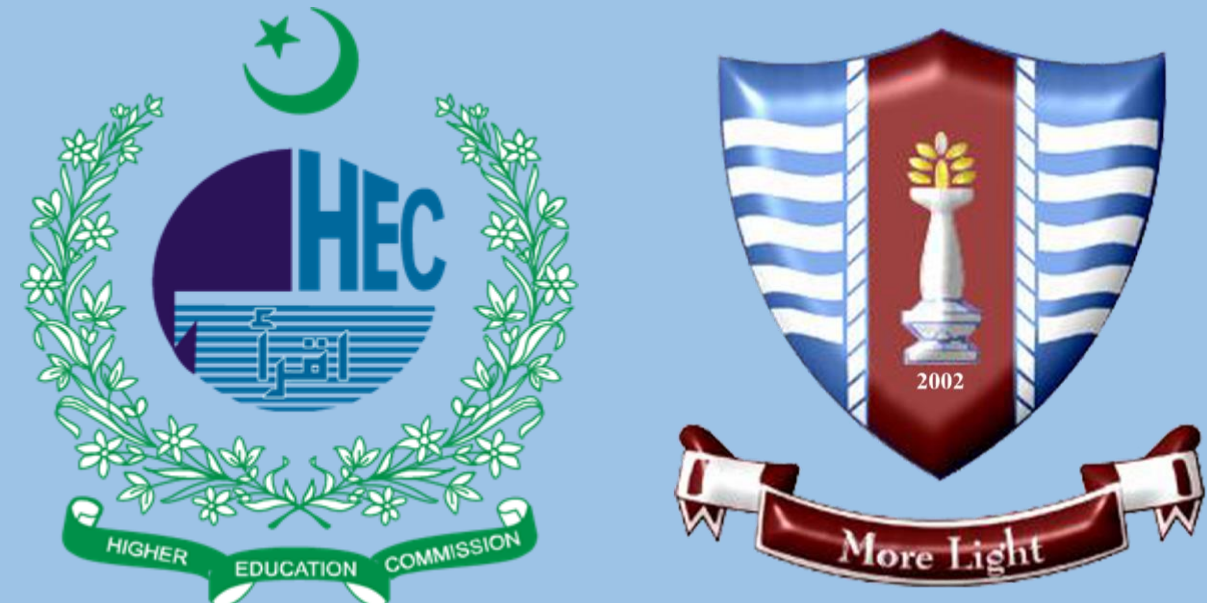




SYNTHESIS, CHARACTERIZATION, THEORETICAL MODELLING, AND BIOLOGICAL ASSESSMENT OF DIORGANOTIN(IV) COMPLEXES WITH THIAZOLE INCORPORATING AZO-SCHIFF BASE LIGANDS

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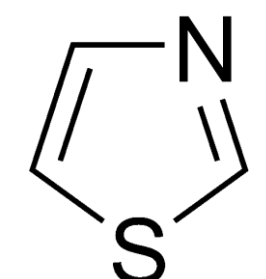
Commonwealth Chemistry
Federation of Chemical Sciences Societies
7th Commonwealth Chemistry Posters
Building Networks to Address the Goals
24-25 June 2026
Poster Code: cwcp2026.031005a

Introduction

Organotin(IV) compounds (R_nSnX_{n-1} , where $n = 1, 2$, or 3) are well known for their diverse biological activities;

- Antimicrobial
- Anti-inflammatory
- Antioxidant
- Anticancer
- Cytotoxic
- Antimalarial
- Antituberculosis
- Antiviral

The incorporation of azo-Schiff base ligands can enhance the stability, reactivity, and bioavailability of organotin complexes, making them attractive candidates for medicinal applications.



While thiazole-containing derivatives are especially notable for their broad pharmacological potential.

Therefore, novel thiazole-based azo-Schiff base ligands (ASbL₁₋₄) were synthesized and successfully coordinated with dimethyltin(IV) and dibutyltin(IV) centers to afford eight new organotin(IV) complexes.

Objective Alignment with SDG 3

This research supports the United Nations Sustainable Development Goal 3 (Good Health and Well-Being) by developing novel therapeutic agents that may help combat infectious diseases and inflammation-related disorders, ultimately promoting improved health outcomes.



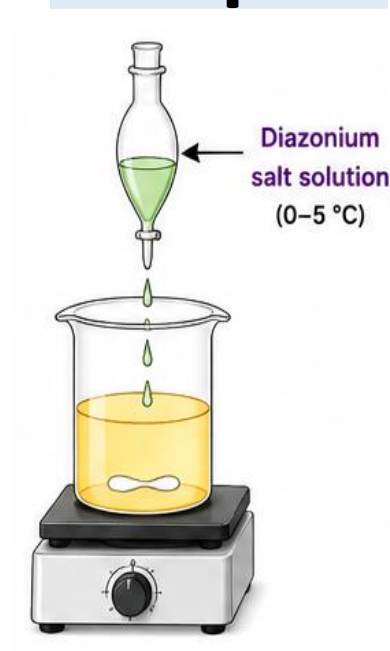
Synthetic Route

Step 1



Schiff base formation from 3,5-dichlorosalicylaldehyde and 2-aminophenol

Step 2



Azo formation from Schiff base via coupling reaction with respective thiazole derivatives

Step 3

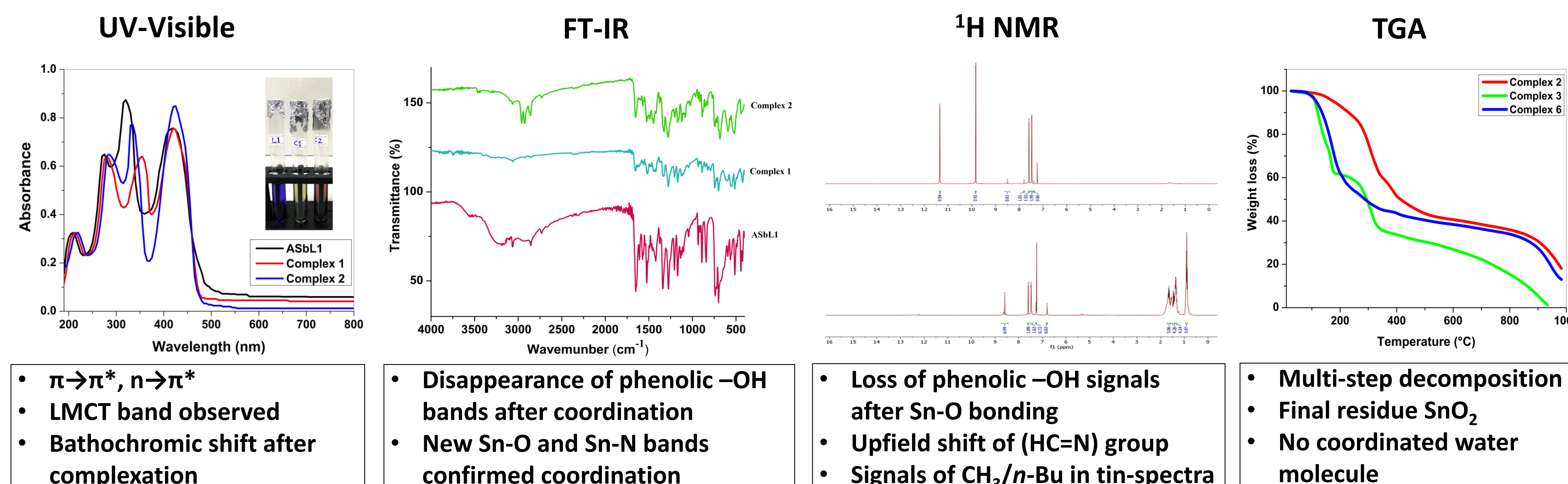


Complex formation by mixing Azo-Schiff base ligand and respective tin(IV) salts

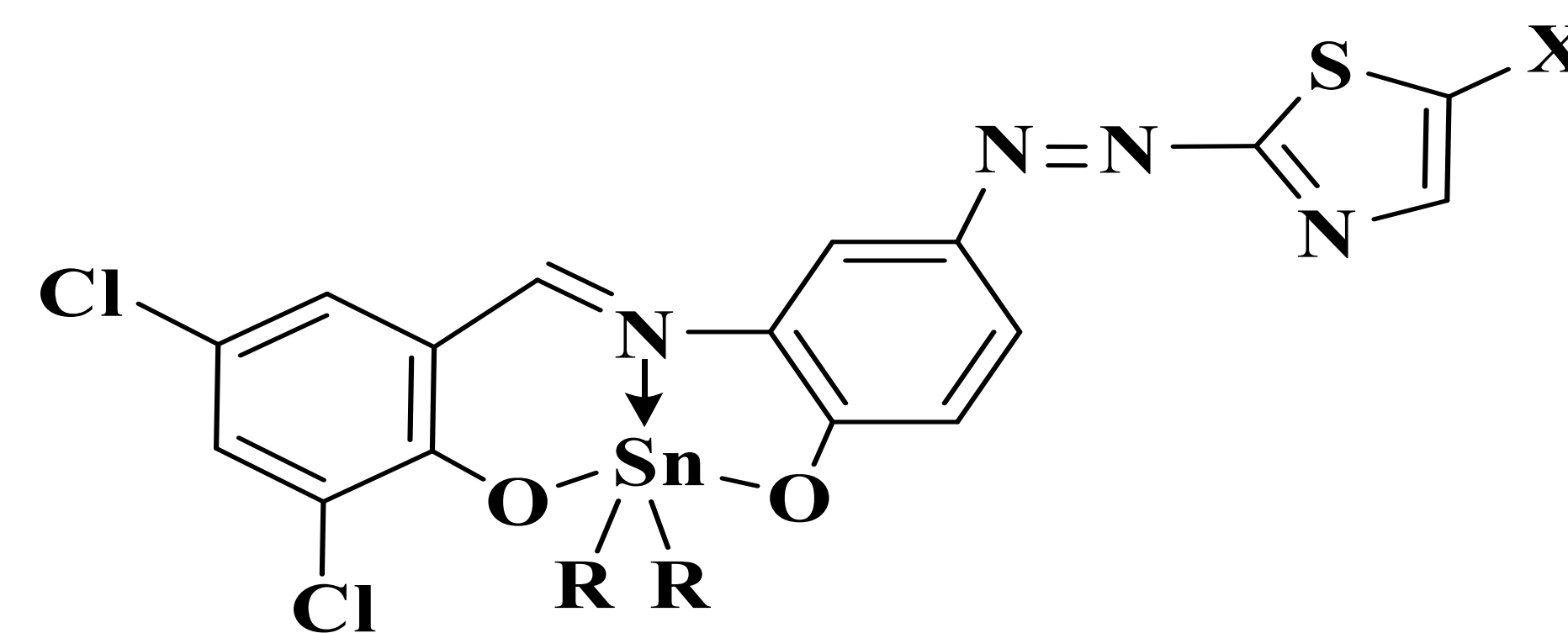
ASbL1 → 2-aminothiazole
ASbL2 → 2-amino-5-nitrothiazole

ASbL3 → 2-aminobenzothiazole
ASbL4 → 2-amino-6-nitrobenzothiazole

Structural Characterizations of tin(IV) complexes

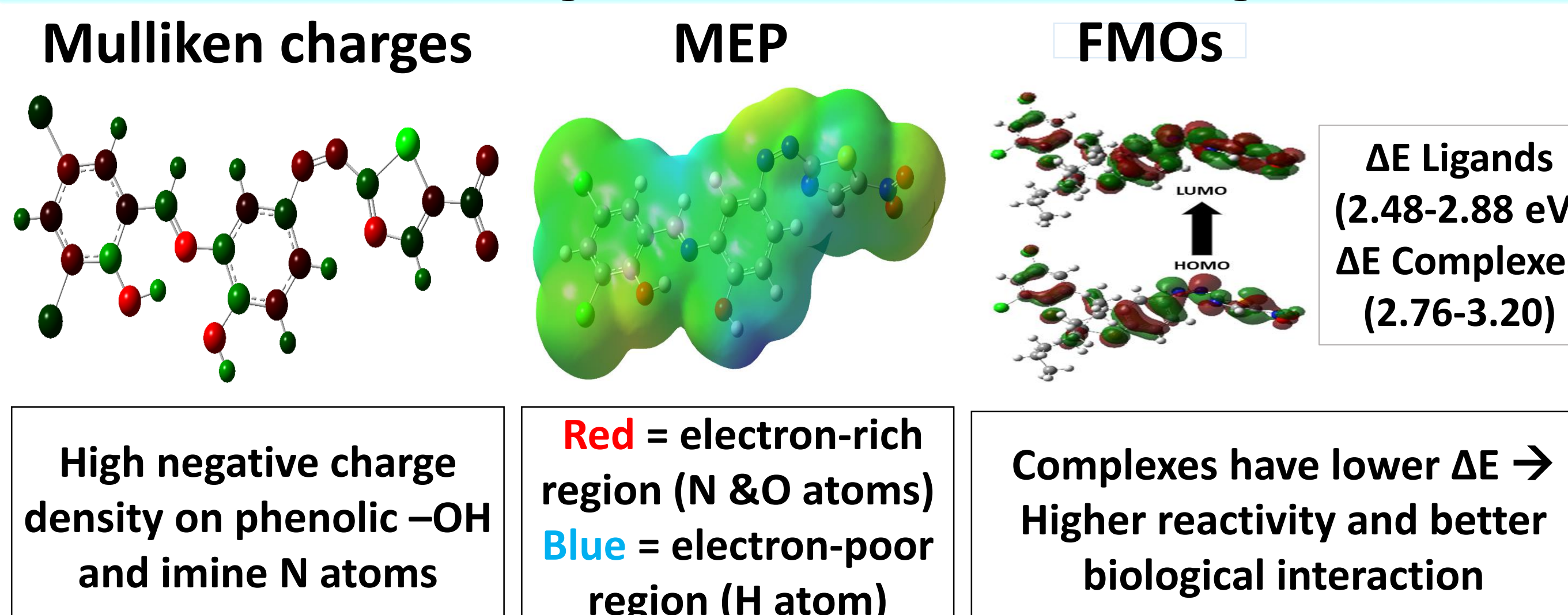


Proposed Coordination Mode



X = H, NO₂
R = Me, *n*-Bu

Density Functional Theory

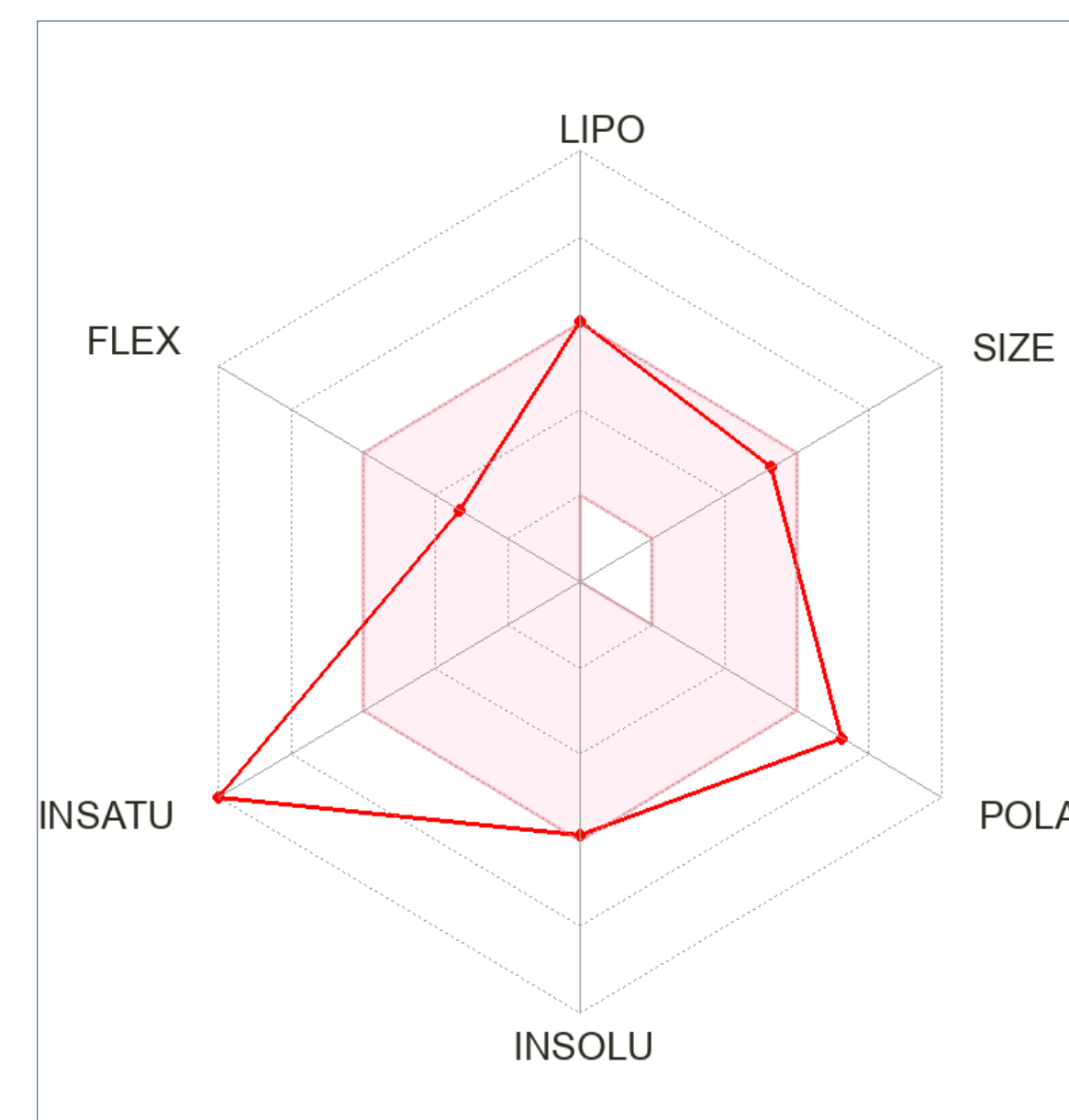


SWISS ADME

Drug-Likeness (Lipinski's Rule)

- Molecular Weight < 500
- H-bond acceptors ≤ 10
- H-bond donors ≤ 5
- Log $p < 5$

Poorly soluble in water (mostly)
Good bioavailability score
NO CYP2D6 inhibition predicted
Acceptable TPSA, molar refractivity,
GI absorption, BBB permeability



Conclusion

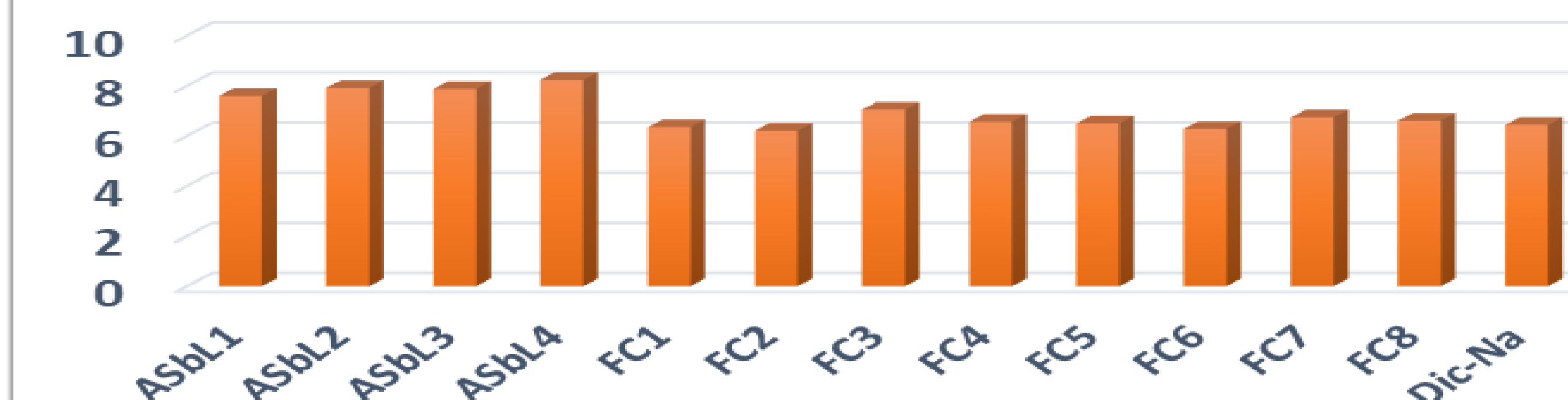
Novel thiazole-based azo ligands and their organotin(IV) complexes were successfully synthesized and characterized by various physicochemical techniques. The organotin(IV) complexes exhibited enhanced biological activities and favorable drug-like properties, highlighting their potential for future therapeutic applications.

References

- P. Barwa, et al., Integrated experimental and theoretical studies of organotin (IV)-Hydrazone complexes: synthesis, structural elucidation, computational insights, and antimicrobial activity, Chem. Biodivers. 22 (9) (2025) e202500544.
- [104] M. Saeed, et al., Synthesis, characterization, in vitro and in silico biological studies of 3d transition metal (II) chelates of a schiff base derived from sulfamethazine and 3, 5-diiodosalicylaldehyde, J. Coord. Chem. 77 (15–16) (2024) 1860–1886.

Biological Activities (Summary)

Anti-Inflammatory Activity (Egg Albumin Method)



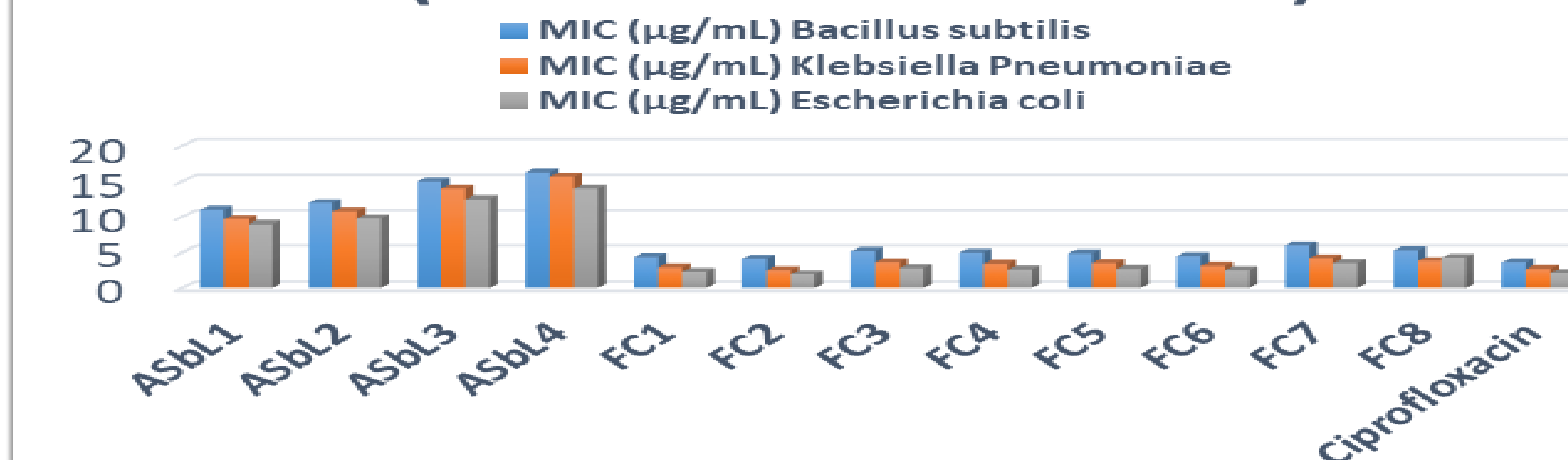
Organotin(IV) Complexes 1, 2, and 4 showed activity better than Sodium Diclofenac (IC₅₀ = 6.44 μ M)

Antioxidant Activity (DPPH assay)



Organotin(IV) Complexes exhibited stronger radical scavenging activity (best Complex 4, IC₅₀ = 3.01 μ M)

Antibacterial Activity (Disc Diffusion Method)



Diorganotin(IV) Complexes > azo Schiff-base ligands
Dibutyltin(IV) complexes > dimethyltin(IV) complexes
More susceptible to Gram-negative bacteria