

Water Pollution Through Arsenic Salts and its Control

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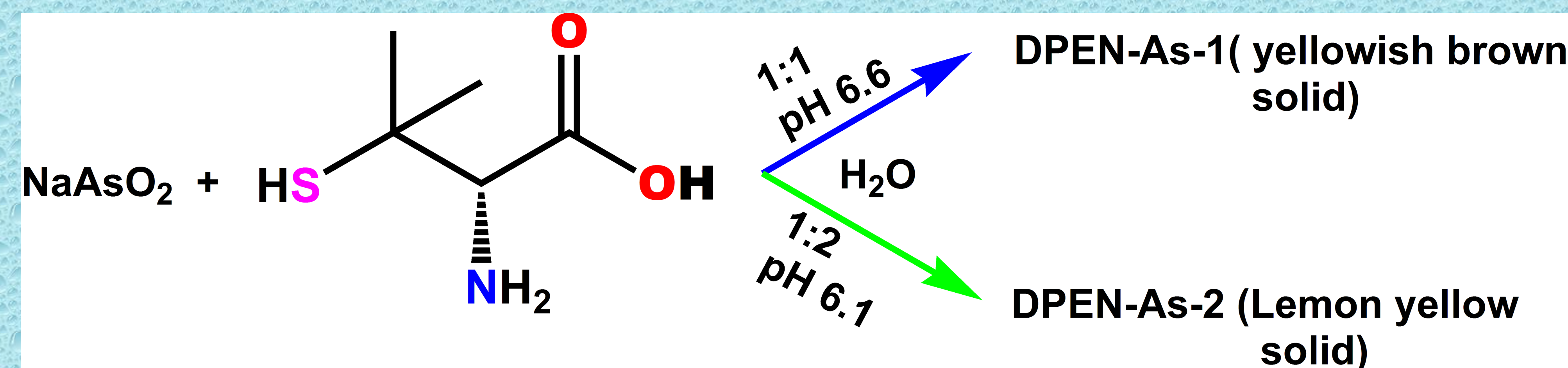
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Introduction:

- Most of the countries in the world depend on ground water as a source for drinking water as well as agricultural purposes
- Literature reveals that ground water, river water and pond water of many countries are contaminated with inorganic arsenic salts as AsO_3^{3-} and AsO_4^{3-} respectively thereby polluting natural water sources
- We are performing interaction of Sodium Arsenite (NaAsO_2) with a drug containing pure D-Penicillamine, a chelating agent with a view to
 - i) Control water pollution through arsenic salts in the laboratory scale and
 - ii) Find out the mode of binding of D-penicillamine having multiple attacking groups like $-\text{SH}$, $-\text{COOH}$ and $-\text{NH}_2$ with sodium arsenite through solid state studies as D-Penicillamine is used as a secondary drug for arsenic poisoning.



Synthetic scheme for the reaction

Important peaks:

DPEN

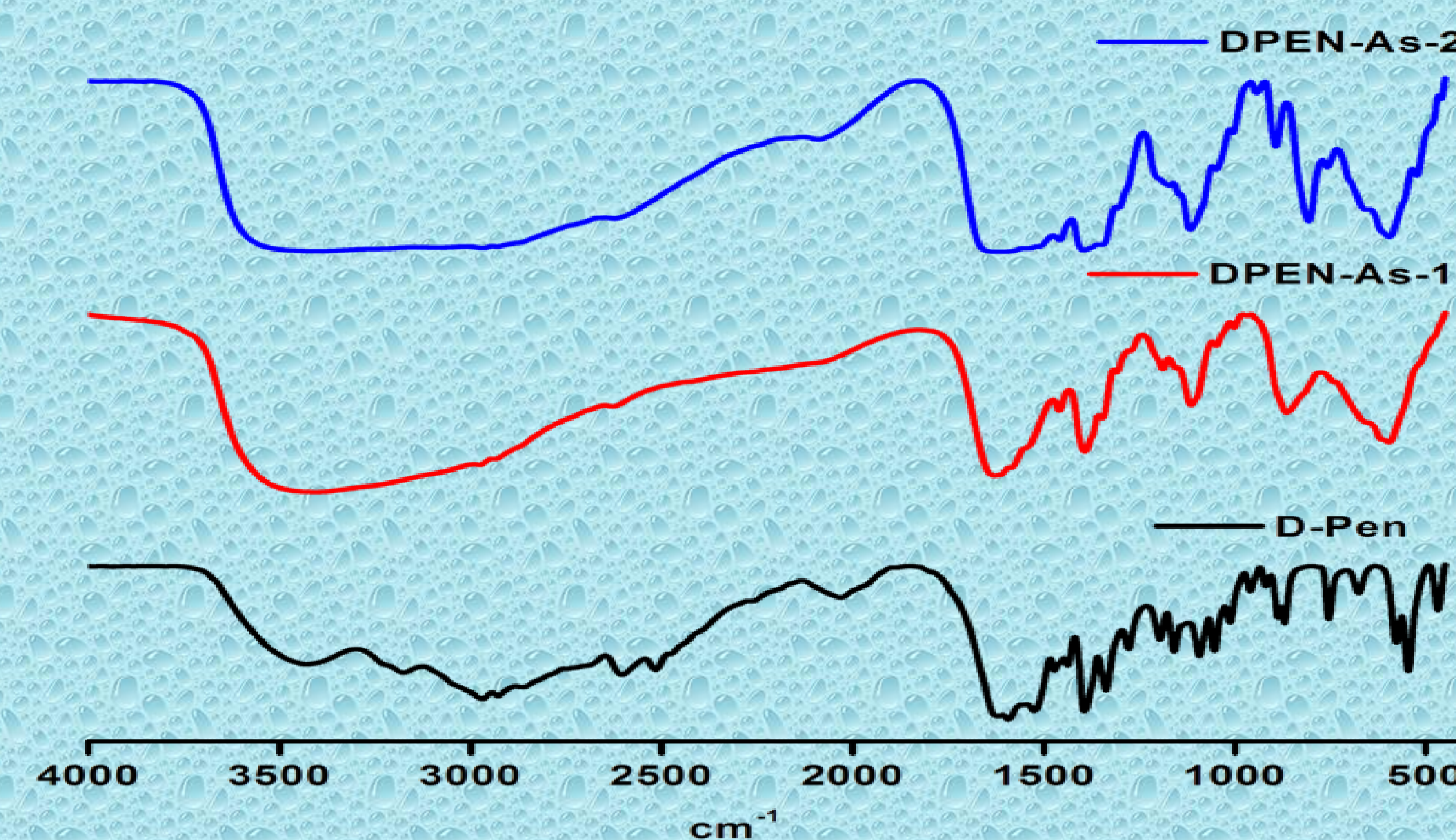
- 3400-3300 $\text{cm}^{-1} \rightarrow -\text{NH}_2$ bending
- 2950-2800 $\text{cm}^{-1} \rightarrow -\text{NH}_3^+$ stretching
- 2500 $\text{cm}^{-1} \rightarrow -\text{SH}$ stretching
- 700-650 $\text{cm}^{-1} \rightarrow -\text{C-S}$ stretching
- 1650-1600 $\text{cm}^{-1} \rightarrow -\text{COO}^-$ stretching

DPEN-As-1/2

- 3500- 3400 $\text{cm}^{-1} \rightarrow -\text{NH}_2$ bending
- 1590-1550 $\text{cm}^{-1} \rightarrow -\text{COO}^-$ stretching
- NH_3^+ stretching vibration is absent in both 1:1 and 1:2 complexes
- $-\text{SH}$ stretching absent

□ Linking of Arsenic to sulphur (thiol) in both the complexes

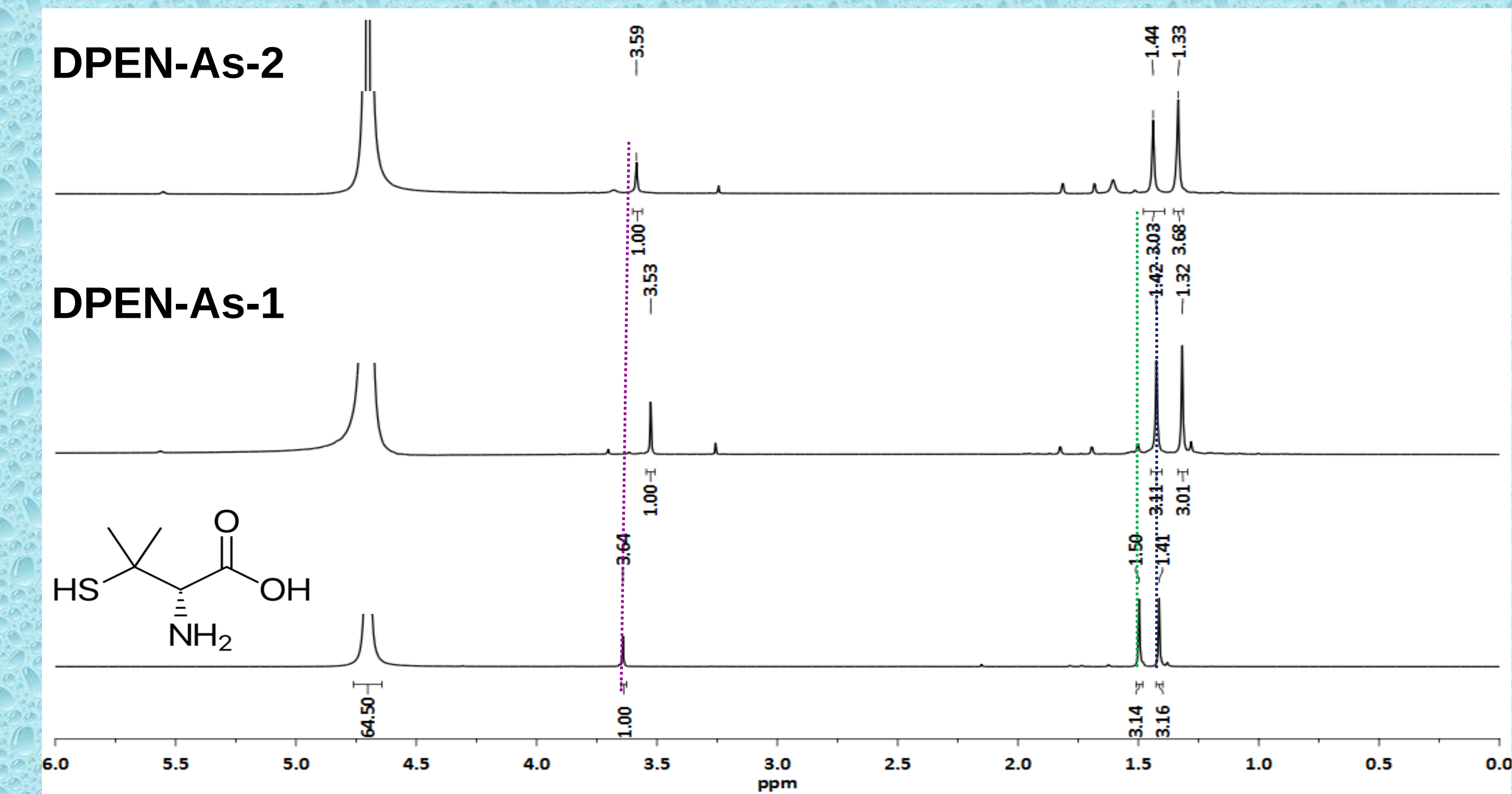
□ Complexes are formed through $-\text{SH}$, $-\text{NH}_2$ and $-\text{COOH}$ linkage.



FTIR Spectra is recorded in Perkin-Elmer Spectrophotometer using KBr pellet in the range of 4400 cm^{-1} – 400 cm^{-1}

References:

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¹H FT-NMR spectra of DPEN and DPEN-As-1/2 complex in D₂O

